

IV. 0.9945 gm. of substance, dried at 100° C., treated with sulphuric and hydrofluoric acids, gave 0.0587 gm. of the chlorides of soda and potash, and 0.0865 gm. of chloroplatinate of potash = 0.0165 gm. of potash and 0.0173 gm. of soda.

Silica,	.	.	.	.	.	.	.	.	.	43.32
Alumina,	.	.	.	.	.	.	.	.	.	13.96
Peroxide of iron,	.	.	.	.	.	.	.	.	.	17.50
Protoxide of iron,	.	.	.	.	.	.	.	.	.	4.36
Lime,	.	.	.	.	.	.	.	.	.	5.96
Magnesia,	.	.	.	.	.	.	.	.	.	5.89
Potash,	.	.	.	.	.	.	.	.	.	1.66
Soda,	.	.	.	.	.	.	.	.	.	1.74
Water,	.	.	.	.	.	.	.	.	.	6.41
										100.80

27. RED CLAY (after removal of carbonate of lime by dilute acid).—Station 286.  
 Lat. 33° 29' S., long. 133° 22' W., 2335 fathoms (Klement).

- I. 1.5318 grms. of substance, dried at 110° C., gave 0.0230 gm. of carbonic acid.
- II. 1.0940 grms. of substance, dried at 110° C., fused with the carbonates of soda and potash, gave 0.0973 gm. of water, 0.4279 gm. of silica, 0.1685 gm. of alumina, 0.1961 gm. of peroxide of iron, 0.0552 gm. of dioxide of manganese, 0.0916 gm. of lime, 0.0720 gm. of pyrophosphate of magnesia.
- III. 0.9345 gm. of substance, dried at 110° C., gave 0.0981 gm. of loss on ignition, and, after being treated with hydrofluoric and sulphuric acids, 0.0434 gm. of the chlorides of soda and potash, and 0.0611 gm. of chloroplatinate of potash.

Silica,	.	.	.	.	.	.	.	.	.	39.10
Alumina,	.	.	.	.	.	.	.	.	.	15.40
Peroxide of iron,	.	.	.	.	.	.	.	.	.	17.93
Manganese dioxide,	.	.	.	.	.	.	.	.	.	5.75
Lime,	.	.	.	.	.	.	.	.	.	8.37
Magnesia,	.	.	.	.	.	.	.	.	.	2.37
Potash,	.	.	.	.	.	.	.	.	.	1.27
Soda,	.	.	.	.	.	.	.	.	.	1.40
Water,	.	.	.	.	.	.	.	.	.	8.89
Carbonic acid,	.	.	.	.	.	.	.	.	.	1.50
										101.98

NOTE.—Before the blow-pipe this substance melted into a deep-coloured scoriaceous bead.

28. RADIOLARIAN OOZE.—Station 265. Lat. 12° 42' N., long. 152° 1' W., 2900 fathoms (Brazier).

		Loss on ignition after drying at 230° Fahr.,	.	4.30
Portion soluble in Hydrochloric Acid = 63.21	}	=	Alumina,	6.75
			Ferric oxide,	11.20
			Calcium phosphate,	0.65
			Manganese oxide,	0.57
			Calcium sulphate,	0.29
			Calcium carbonate,	2.54
			Magnesium carbonate,	2.46
Portion insoluble in Hydrochloric Acid = 32.49	}	=	Silica,	38.75
			Alumina,	6.19
			Ferric oxide,	3.09
			Lime,	1.85
			Magnesia,	0.34
		Silica,	21.02	
				100.00